# **1. MEMBRANE TRANSPORT**

- small soluble molecules;
- inorganic ions and small water-soluble molecules;

1

- large particles.

# **2. CELL JUNCTIONS**

# **MEMBRANE TRANSPORT**

I. small soluble molecules (hydrophobic/nonpolar) -<u>without</u> specialized transmembrane proteins (simple DIFFUSION)

II. inorganic ions and small water-soluble molecules – <u>with</u> specialized transmembrane proteins:

> - "carrier" p.-moving parts (active/passive); Energy source - ATP/ion gradients;

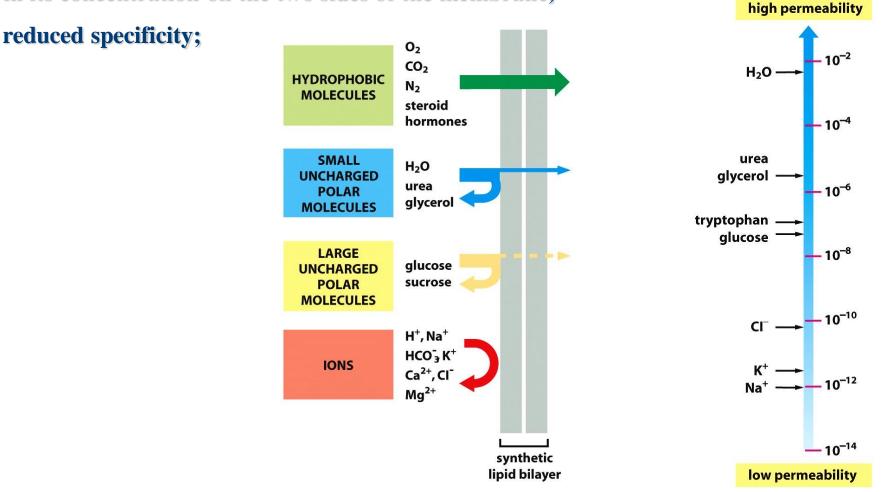
- "channel" p .--form a narrow hydrophilic pore (passive - facilitate diffussion);

**III.** large particles <= vesicules

- endocytosis;
- exocytosis;

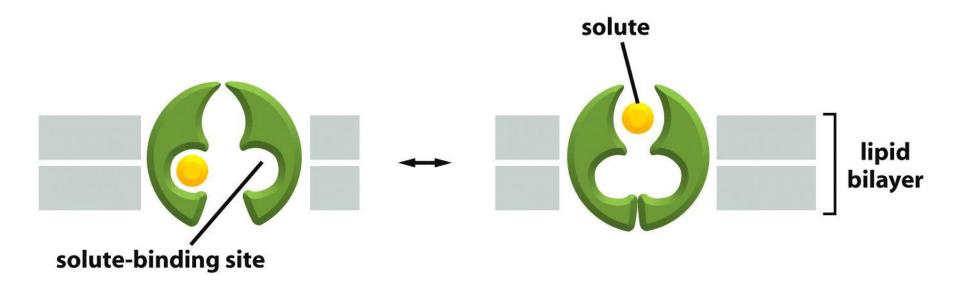
# I. <u>simple DIFFUSION</u> across a protein-free lipid bilayer down its concentration gradient

the rate of flow of a solute across the bilayer is directly proportional to the difference in its concentration on the two sides of the membrane;



### II. A. <u>TRANSPORTERS</u> (Carriers/Permeases)

undergo a series of <u>reversible conformational changes</u> to transfer the bound solute across the membrane.



#### TRANSPORTER

Figure 11-3a Molecular Biology of the Cell (© Garland Science 2008)

### **II.A. TRANSPORTERS (Carriers/Permeases)**

#### A. +/- energy source - transport

- active ("uphill") "pumps" (ATP/ion gradient)
- passive ("downhill") passive transport, or facilitated diffusion, concentration gradient;

#### **B.** Systems

- uniport: one type of molecule;

Antiport – Exchangers

#### A. +/- energy source - transport

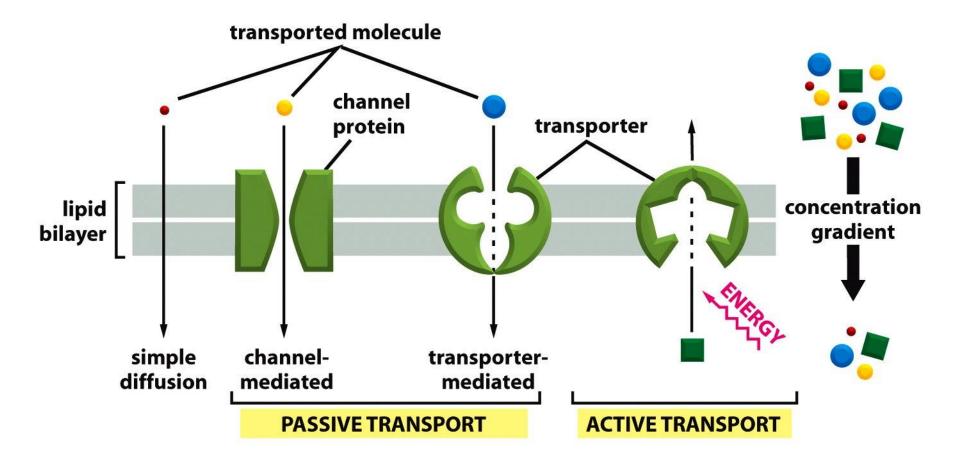
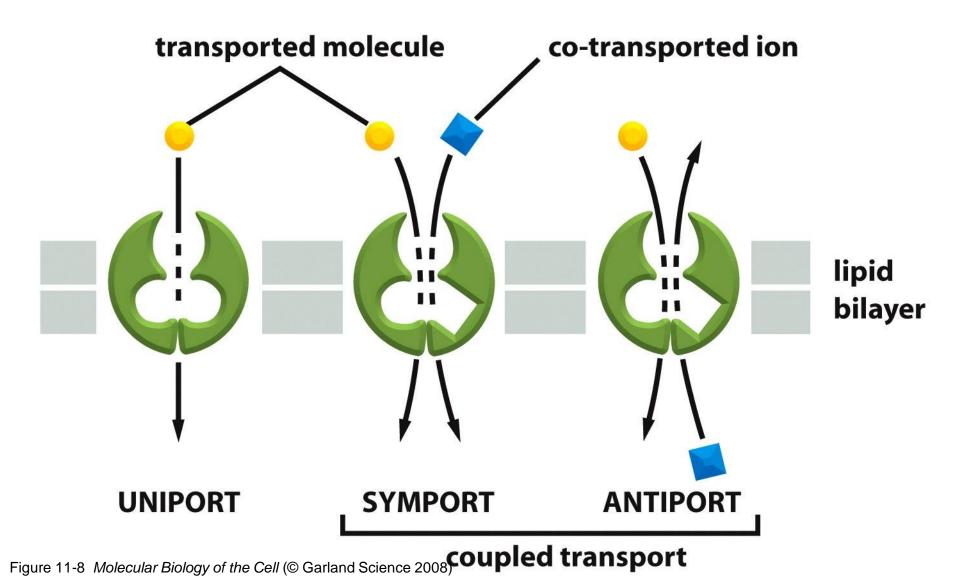


Figure 11-4a Molecular Biology of the Cell (© Garland Science 2008)

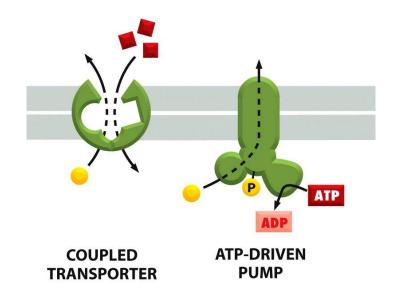
#### **B. Uniport/coupled**

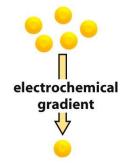


# II.A. TRANSPORTERS (Carriers/Permeases) ACTIVE MEMBRANE TRANSPORT

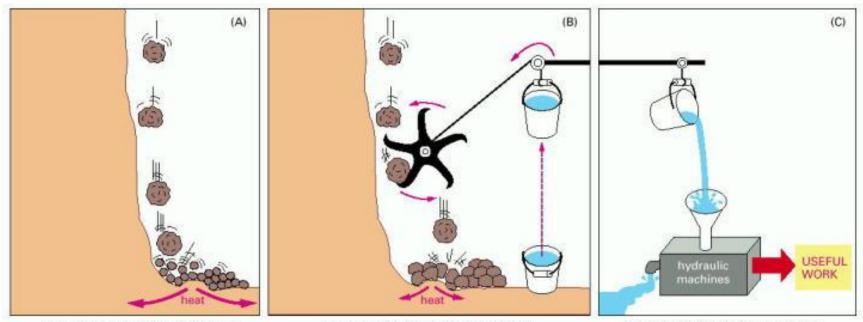
1. <u>Coupled transporters</u> couple the uphill transport of one solute across the membrane to the downhill transport of another;

2. <u>ATP-driven pumps</u> couple uphill transport to the hydrolysis of ATP





# **Energy conversion**



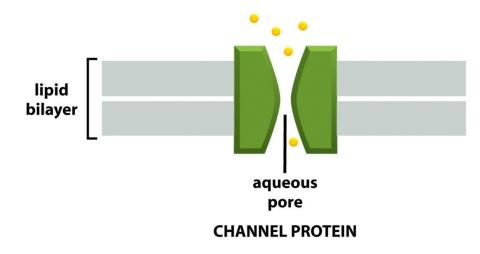
kinetic energy transformed into heat energy only

part of the kinetic energy is used to lift a bucket of water, and a correspondingly smaller amount is transformed into heat

the potential kinetic energy stored in the raised bucket of water can be used to drive hydraulic machines that carry out a variety of useful tasks

# II. B. <u>ION CHANNELS</u>

- interact with the solute to be transported much more weakly;
- form aqueous pores that extend across the lipid bilayer;
- when open, these pores allow specific solutes (usually inorganic ions of appropriate size and charge) to pass through them and thereby cross the membrane;
- one class of channel proteins found in virtually all animals forms gap junctions between two adjacent cells



# II. B. ION CHANNELS

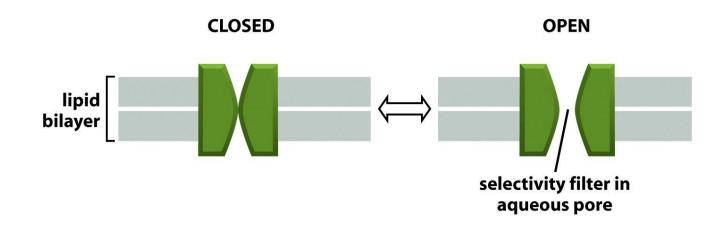
- transport through channels occurs at a much faster rate - 100 milion ions/seconds  $-10^5$  faster than a carrier protein;
- higly selective pores primarily Na+, K+, Ca2+, or Cl-;
- can close and open rapidly;
- ALWAYS PASSIVE (downhill)!!!
- Fluctuate between open and closed states.

# II. B. IONS CHANNELS

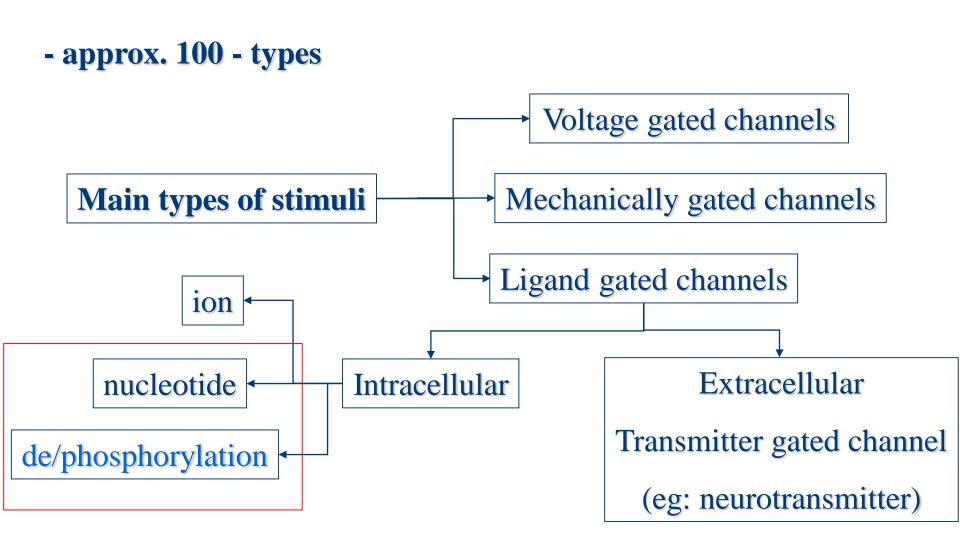
### **Ion-Selective and Fluctuate Between Open and Closed States**

#### ION CHANNELS vs SIMPLE AQUEOUS CHANNELS

- only ions of appropriate size and charge can pass (selectivity filter);
- ion channels are not continuously open, they are gated;
- the gate opens in response to a specific stimulus.



#### II. B. IONS CHANNELS



# II. B. IONS CHANNELS

PROLONGED (chemical or electrical) STIMULATION, most channels go into a closed "desensitized" or "inactivated" state, in which they are REFRACTORY to further opening UNTIL THE STIMULUS HAS BEEN REMOVED.

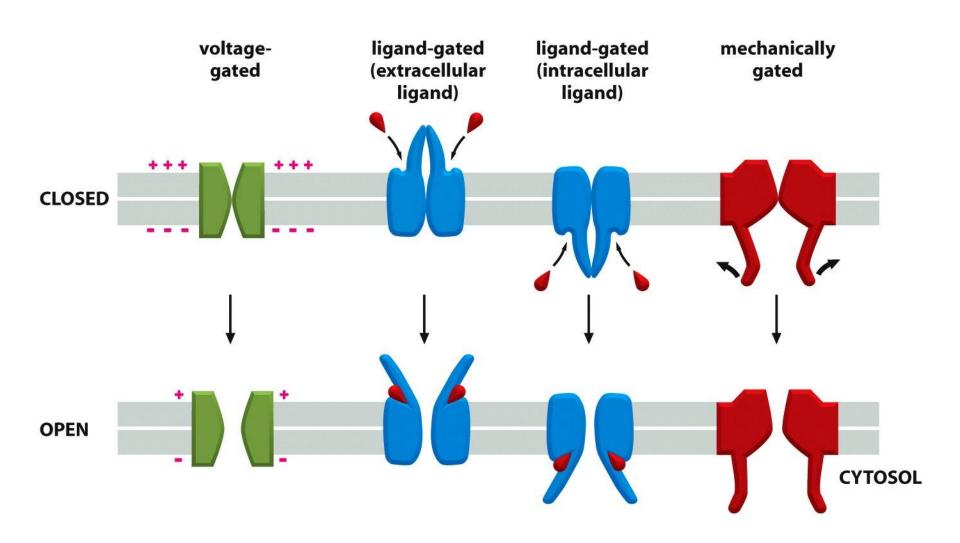
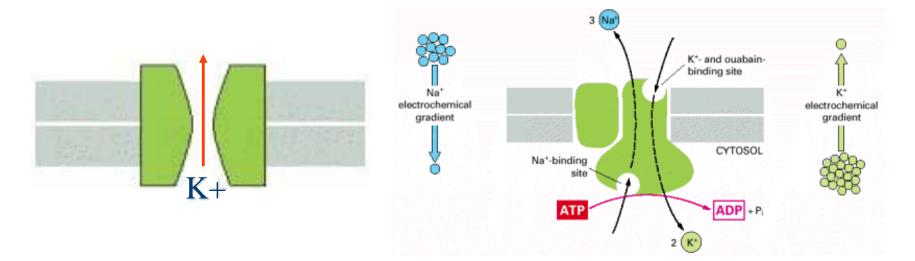


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# K+ leak channels

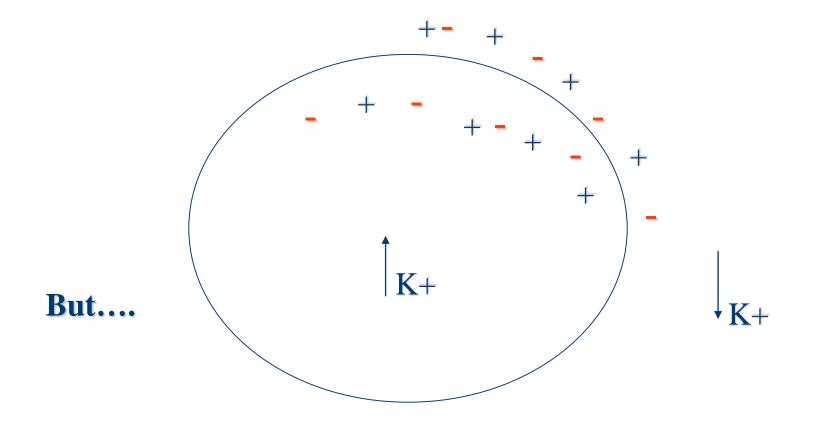
#### no stimuli required!

### balancing role => membrane potential across all plasma membranes;



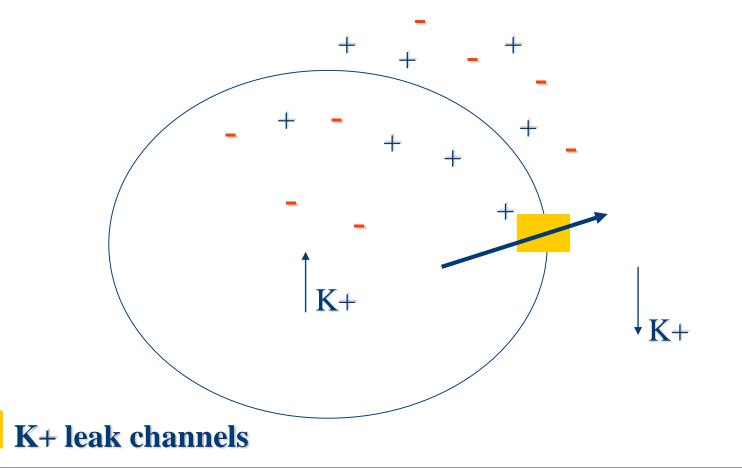
**K+ transport across membrane – IDEAL CELL** 

I. If no voltage gradient across the plasma membrane (the membrane potential is zero);



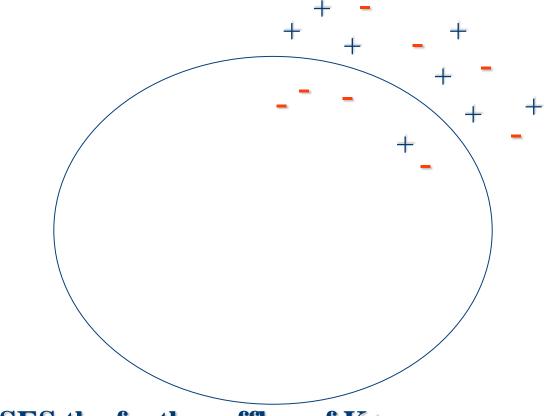
### K+ transport across membrane

**II.** K+ will tend to leave the cell through the K+ leak channels, driven by its concentration gradient.



K+ transport across membrane

=> K+ leaves behind an unbalanced negative charge <=> electrical field/ membrane potential



**OPPOSES the further efflux of K+** 

|        | _           | +           | -      | +           | -           | +      | +           | -                | +           | -      | +                | -                | +                |
|--------|-------------|-------------|--------|-------------|-------------|--------|-------------|------------------|-------------|--------|------------------|------------------|------------------|
| -      | +           | -           | +      | —           | +           | -      |             | +                | _           | +      | -                | +                | -                |
| +      | -           | +           | -      | +           |             | +      | +           |                  | +           | -      | +                | _                | +                |
| _      | +           | —           | +      | -           | +           |        | -           | +                | —           | +      | _                | +                | -                |
| +      | -           | +           | -      | +           | -           | +      | +           | _                | +           | -      | +                | _                | +                |
| _      | +           | _           | +      | -           | +           | -      | -0          | +                | -           | +      | -                | +                | -                |
| +      | _           | -           |        |             |             |        | +           |                  |             |        |                  |                  | 0.00             |
|        |             | т           |        | Τ.          |             | T      | -           |                  | Ŧ           |        | +                | -                |                  |
| _      | +           | <i>.</i> 7  |        | 107         |             |        | 1.00        | +                | 7.0         | +      | 10               |                  | 93 <b>7</b> 6    |
| -<br>+ |             | -           |        | -           | +           | -      | 1.00        | +                | _           |        | -                | +                | -                |
| -<br>+ | +           | -<br>+      | +<br>- | -<br>+      | +<br>-      | -<br>+ | _           | +                | -<br>+      | +      | -<br>+           | +                | -<br>+           |
| -<br>+ | +<br>-<br>+ | -<br>+<br>- | +<br>- | -<br>+<br>- | +<br>-<br>+ | +      | -<br>+<br>- | +<br>-<br>+<br>- | -<br>+<br>- | +<br>- | -<br>+<br>-<br>+ | +<br>-<br>+<br>- | -<br>+<br>-<br>+ |

exact balance of charges on each side of the membrane; membrane potential = 0

| + - +   | - +               | - +               |     |         | +           | -           | +             | -           |
|---------|-------------------|-------------------|-----|---------|-------------|-------------|---------------|-------------|
| - + -   | + -               | + +               |     | - +     |             | +           | -             | +           |
| + - +   | - +               | - +               |     |         | +           | -           | +             | -           |
| - + -   | + -               | + +               |     | - +     | -           | +           | -             | +           |
| + - +   | - +               | -+                | - 4 | + -     | +           | -           | +             | s <b></b> - |
| - + -   | + -               | + +               |     | - +     | -           | +           | -             | +           |
|         |                   |                   |     |         |             |             |               |             |
| + - +   | - +               | - +               | - 4 |         | +           | _           | +             | :           |
| + - +   | 737A              |                   |     | <br>- + | 0.5         |             | 3 <b>5</b> 40 |             |
| 121     | + -               | ++                |     |         | _           | +           | _             | +           |
| - + -   | + -<br>- +        | + +<br>- +        |     | - +     | -<br>+      | +<br>-      | -<br>+        | +<br>-      |
| - + - + | + -<br>- +<br>+ - | + +<br>- +<br>+ + |     | - +<br> | -<br>+<br>- | +<br>-<br>+ | -<br>+<br>-   | +<br>-<br>+ |

a few of the positive ions (*red*) cross the membrane from right to left, leaving their negative counterions (*red*) behind; this sets up a nonzero membrane potential



=> <u>the efflux of K+ STOPS</u> when the membrane potential reaches a value at which this <u>electrical driving force on K+</u> <u>exactly balances the effect of its concentration gradient</u>

<=> ELECTROCHEMICAL GRADIENT FOR K+ = 0

no net flow of ions across the plasma membrane <=> RESTING MEMBRANE POTENTIAL

# **MEMBRANE POTENTIAL**

- <u>Voltage difference</u> across a membrane due to a <u>slight excess of</u> positive ions on one side and of negative ions on the other;

- <u>The charge differences</u> result both from <u>active electrogenic</u> <u>pumping</u> and from <u>passive ion diffusion</u>.

- <u>Na+ /K+ pump</u> helps <u>maintain an osmotic balance</u> across the animal cell membrane by keeping the intracellular concentration of Na+ low.

=> other cations have to be plentiful there to balance the charge carried by the cell's fixed anions—the negatively charged organic molecules that are confined inside the cell.

### K+ transport across membrane

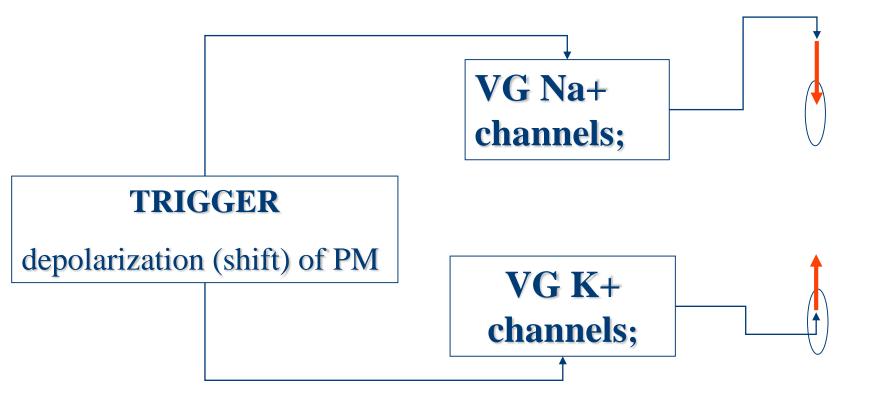
actively pumped into the cell by the Na+/K+ pump

can also move freely in or out through the K + leak channels in the plasma membrane

- 2 types of forces to move K+ => K+ to equilibrium:
- the chemical gradient (high inside/ low outside);
- electrical force exerted by an excess of negative charges attracting K+ into the cell

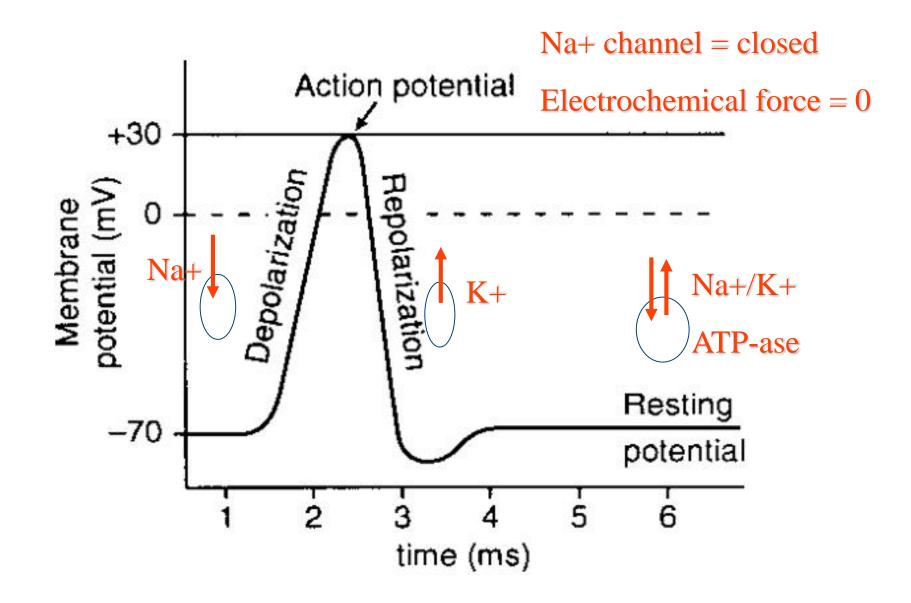
### **Voltage-gated (VG) cation channels**

-responsible for generating the action potentials;



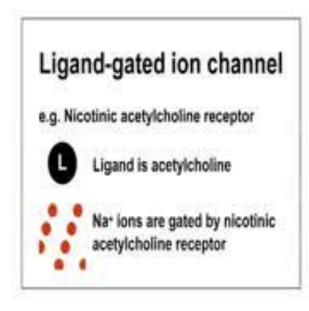
#### **ACTION POTENTIAL**

- is a short-lasting event electrical membrane potential of a cell rapidly rises and falls, following a consistent trajectory;
- occurs in several types of animal cells called excitable cells:
  - neuron cell-to-cell communication
  - muscle cells Ist step in the chain of events contraction
- endocrine cells (eg: beta cells of the pancreas- provoke release of insulin);



# **Ligand-gated channels**

Ligand gated ion channel - structure and function



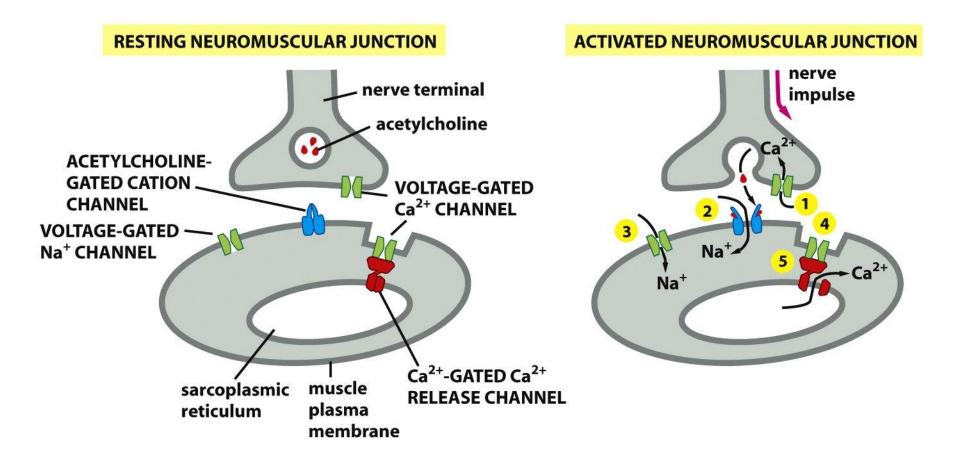
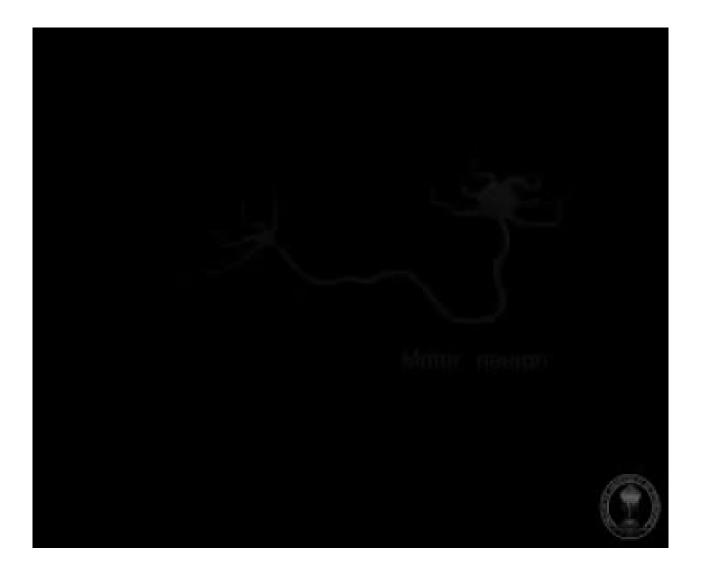
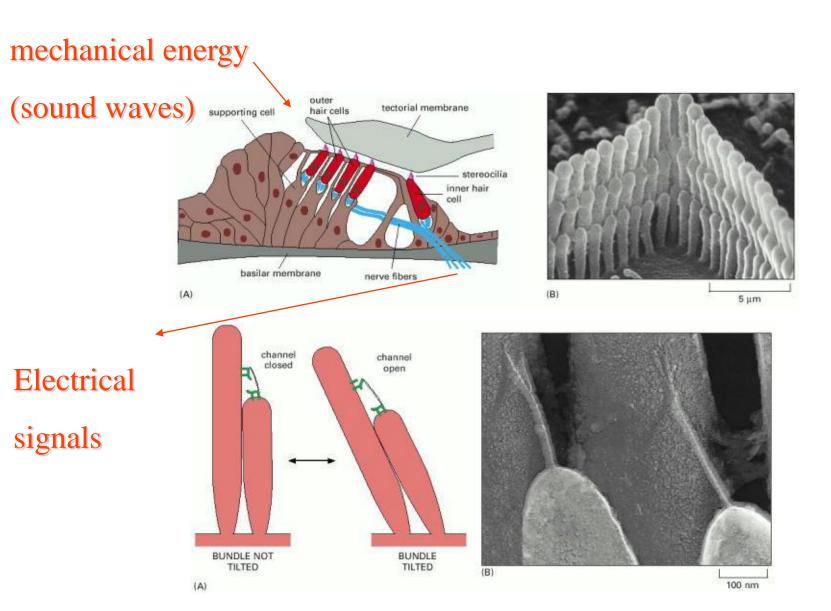


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# Motor unit & ion channels



# **Mechanically gated channels**



# **Mechanically gated channels – K+ channels**

